

Balancing Chemical Equations Gizmo Answer Key

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Balancing Chemical Equations Gizmo Answer

Gizmo: Balancing Chemical Equations

1 Match: Most chemical reactions can be classified as one of four types Using the chemical equations in the Gizmo as a guide, match the following definitions to the type of reaction Using the Gizmo, match the descriptions below to the correct type of reaction ___ One reactant is broken down into two or more products A Synthesis

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Student Exploration: Balancing Chemical Equations

Gizmo Warm-up In a chemical reaction, reactants interact to form products This process is summarized by a chemical equation In the Balancing Chemical Equations Gizmo™, look at the floating molecules below the initial reaction: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ 1 How many atoms are in a hydrogen molecule (H_2)? ___ 2 How many atoms are in an oxygen molecule (O_2)

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Balancing Chemical Equations Answer Key

Balancing Chemical Equations - Answer Key Balance the equations below: 1) $1 \text{ N}_2 + 3 \text{ H}_2 \rightarrow 2 \text{ NH}_3$ 2) $2 \text{ KClO}_3 \rightarrow 2 \text{ KCl} + 3 \text{ O}_2$ 3) $2 \text{ NaCl} + 1 \text{ F}_2 \rightarrow 2 \text{ NaF} + 1 \text{ Cl}_2$ 4) $2 \text{ H}_2 + 1 \text{ O}_2 \rightarrow 2 \text{ H}_2\text{O}$ 5) $1 \text{ Pb(OH)}_2 + 2 \text{ HCl} \rightarrow 2 \text{ H}_2\text{O} + 1 \text{ PbCl}_2$ 6) $2 \text{ AlBr}_3 + 3 \text{ K}_2\text{SO}_4 \rightarrow 6 \text{ KBr} + 1 \text{ Al}_2(\text{SO}_4)_3$

Balancing Chemical Equations - AP Chemistry

Balancing Chemical Equations - Answer Key Balance the equations below: 1) $1 \text{ N}_2 + 3 \text{ H}_2 \rightarrow 2 \text{ NH}_3$ 2) $2 \text{ KClO}_3 \rightarrow 2 \text{ KCl} + 3 \text{ O}_2$ 3) $2 \text{ NaCl} + 1 \text{ F}_2 \rightarrow 2 \text{ NaF} + 1 \text{ Cl}_2$ 4) $2 \text{ H}_2 + 1 \text{ O}_2 \rightarrow 2 \text{ H}_2\text{O}$ 5) $1 \text{ Pb(OH)}_2 + 2 \text{ HCl} \rightarrow 2 \text{ H}_2\text{O} + 1 \text{ PbCl}_2$ 6) $2 \text{ AlBr}_3 + 3 \text{ K}_2\text{SO}_4 \rightarrow 6 \text{ KBr} + 1 \text{ Al}_2(\text{SO}_4)_3$ 7) $1 \text{ CH}_4 + 2 \text{ O}_2 \rightarrow 1 \text{ CO}_2 + 2 \text{ H}_2\text{O}$ 8) $1 \text{ C}_3\text{H}_8 + 5 \text{ O}_2 \rightarrow 3 \text{ CO}_2 + 4 \text{ H}_2\text{O}$ 9) $2 \text{ C}_8\text{H}_{18} + 25 \text{ O}_2 \rightarrow 16 \text{ CO}_2 + 18 \text{ H}_2\text{O}$ 10) $1 \text{ FeCl}_3 + 3 \dots$

Student Exploration: Chemical Equations

Balancing equations Get the Gizmo ready: Erase the chemical formulas in each text box Introduction: In a chemical reaction, the reactants are the substances that enter into the reaction, and the products are the substances that are made in the reaction A chemical

Chemical Equations Gizmo Answers Online

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Balancing Equations: Practice Problems - North Allegheny

Balancing Equations: Answers to Practice Problems 1 Balanced equations (Coefficients equal to one (1) do not need to be shown in your answers)

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Balancing Chemical Equations - teachnlearnchem.com

KEY Chemistry: Balancing Chemical Equations Directions: First, balance each of the chemical equations below Then, classify each reaction as synthesis, decomposition, single-replacement, or double-replacement To earn full credit, write the words out

Gizmo Chemical Equations Answers - smartdesigninteriors.ca

Explore Learning Gizmo Answer Key Balancing Chemical Equations This is the fourth lesson in the chemical reaction unit that covers HS-PS1-7, the uses of mathematical representations to support the claim that atoms, and therefore mass, are conserved during a chemical reaction; and DCI-PS1B, the fact that atoms are conserved, together with knowledge of the chemical properties of the elements

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Lesson Info: Chemical Equations Gizmo | ExploreLearning www.explorelearning.com > Gizmos Chemical Equations Practice balancing chemical equations by changing the coefficients of reactants and products As the equation is manipulated, the amount of $\hat{\epsilon}$ | Balancing Chemical Equations - ...

Student Exploration: Chemical Equations - PAP

Student Exploration: Chemical Equations - PAP Login to www.explorelearning.com to complete this assignment Define the following terms on a separate paper by 10/1: Coefficient, Conservation of matter, Molecule, Product, Reactant, Subscript Prior Knowledge Questions (Do these BEFORE using the Gizmo) 1 A candle is placed on one pan of a balance

Balancing Equations Date: Name - Mr. Arthur's Science Page

Balancing Equations Balance the following chemical equations
 1 $2 \text{Fe} + 3 \text{H}_2\text{SO}_4 \rightarrow \text{Fe}_2(\text{SO}_4)_3 + 3 \text{H}_2$
 2 $2 \text{C}_2\text{H}_5 + 7 \text{O}_2 \rightarrow 6 \text{H}_2\text{O} + 4 \text{CO}_2$
 3 $3 \text{KOH} + 1 \text{H}_3\text{PO}_4 \rightarrow 1 \text{K}_3\text{PO}_4 + 3 \text{H}_2\text{O}$
 4 $1 \text{SnO}_2 + 2 \text{H}_2 \rightarrow 1 \text{Sn} + 2 \text{H}_2\text{O}$
 5 $4 \text{NH}_3 + 5 \text{O}_2 \rightarrow 4 \text{NO} + 6 \text{H}_2\text{O}$
 6 $2 \text{KNO}_3 + 1 \text{H}_2\text{CO}_3 \rightarrow 1 \text{K}_2\text{CO}_3 + 2 \text{FINO}_3$
 7 $1 \text{B}_2\text{Br}_6 + 6 \text{HNO}_3 \rightarrow 2 \text{B}(\text{NO}_3)_3 + 6 \text{HBr}$
 8 $2 \text{BF}_3 + 3 \text{Li}_2\text{SO}_3 \rightarrow 1 \text{B}_2(\text{SO}_3)_3 + 6 \text{LiF}$
 9 $4 (\text{NH}_4)$